C4	0.2707(1)	-0.0812	(1)	-0.1874(1)	0.060 (1
C5	0.2714(1)	-0.1584	(1)	-0.0851(2)	0.053 (1
C6	0.2541 (1)	0.1414	(1)	0.0840(1)	0.042 (1)
C7	0.1934(1)	-0.0655	(1)	0.1140(1)	0.039 (1
C8	0.1405 (1)	-0.0746	(1)	0.2637 (2)	0.051 (1)
C9	0.2041 (1)	-0.0820	(1)	0.3912(2)	0.053 (1
C10	0.2409(1)	-0.0058	(1)	0.4079(1)	0.052 (1)
C11	0.2016(1)	0.0545	(1)	0.2922 (2)	0.046 (1)
C12	0.2333 (1)	0.0272	(1)	0.1313(1)	0.035 (1)
C13	0.1096(1)	0.0188	(1)	0.2872 (2)	0.058 (1)
01	0.3826(1)	0.0394	(1)	0.2321 (1)	0.044 (1)
Т	able 2. Sele	ected geon	netric	parameter	s (Å, °)
C1C2		1.539 (2)	C1-	C12	1.542 (1
C101		1.423(1)	C2-	C3	1.499 (2
C3C4		1.315 (3)	C4	C5	1.503 (2
C5-C6		1.544 (2)	C6-	C7	1.531 (2
C7C8		1.565 (2)	C7-	C12	1.574 (2
C8C9		1.504 (2)	C8	C13	1.540 (2
C9C10		1.323 (2)	C10	C11	1.515 (2
C11C12	2	1.566 (2)	C11	C13	1.533 (2
01-01		2.736 (1)			
01C1	C12	114.3 (1)	01-	C1C2	108.9 (1
C12C1-	C2	116.6 (1)	C3	C2C1	114.7 (1
C4—C3	-C2	123.5 (1)	C5–	C4C3	125.7 (1
C6C5	C4	116.5 (1)	C7–	-C6C5	114.0 (1
C12-C7-		101.9 (1)	C12	C7C6	118.5 (1)
C8C7		113.6 (1)	C13	C8C9	100.1 (1
C13C8-	C 7	101.2(1)	C9	C8C7	107.4 (1
C10-C9	C8	107.4 (1)	C11	C10C9	107.6 (1
C13—C1	1—C12	99.7 (1)	C13	C11C10	99.9 (1
C12C1	1—C10	108.8 (1)	C11	C12C7	102.3 (1
CII—CI	2—C1	112.9 (1)	C7	C12C1	124.5 (1
C11-C1	3-68	93.1 (1)			

Table 3. Comparison of selected bond lengths (Å) and bond angles (°) of known crystal structures

	C7—C12	C9-C10	C1-C12-C11	C6-C7-C8
(I)	1.574 (2)	1.323 (2)	112.9 (1)	113.6(1)
(II)	1.571 (2)	1.318 (2)	110.8 (1)	113.6(1)
(III) (molecule 1)	1.564 (3)	1.307 (4)	112.4 (2)	112.9 (2)
(molecule 2)	1.561 (4)	1.313 (5)	113.1 (2)	112.2 (3)
(IV)	1.569 (2)	1.312 (2)	112.3 (2)	111.8 (1)

Table 4. Comparison of r.m.s. deviations (Å) of the C_{13} skeleton of tricyclo[8.2.1.0^{2,9}]trideca-5,11-diene in known crystal structures

R.m.s. deviation	(I) 0.09	(II) 0.17	(III) 0.09	(IV) 0.21	

Data collection: CAD-4 Software (Enraf-Nonius, 1989). Cell refinement: CAD-4 Software. Data reduction: DATAP (Coppens, Leiserowitz & Rabinovitch, 1965). Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993), GFMLX (Flack, 1983). Molecular graphics: SHELXTL-Plus (Sheldrick, 1989), XANADU (Roberts & Sheldrick, 1975), SYBIL (Tripos Associates Inc., 1994). Software used to prepare material for publication: DAESD (Davis & Harris, 1970).

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N-Benzoyl-2-phenyldecahydroquinolin-4one, $C_{22}H_{23}NO_2$

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Abstract

The title molecule consists of a piperidin-4-one ring fused to a cyclohexane moiety. The piperidine ring adopts a twisted conformation and the cyclohexane ring adopts a chair conformation. The crystal structure determination confirms the *trans* fusion of the two rings.

† Deceased.

Lists of structure factors, anisotropic displacement parameters, Hatom coordinates and complete geometry have been deposited with the IUCr (Reference: NA1181). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

C22H23NO2

Comment

The title compound. (I), has been analysed as part of our crystallographic studies on substituted decahydroquinolines.



Fig. 1 shows a perspective view of the molecule with the atomic numbering (ORTEPII; Johnson, 1976). The H9—C9—C10—H10 torsion angle is $-173.3 (3)^{\circ}$. This confirms the trans configuration of the title compound. The departures of angles C2-N1-C11 [121.1 (3)°], C2-N1-C9 [117.8 (3)°] and C9-N1-C11 $[116.9(2)^{\circ}]$ from tetrahedral values indicate a flattening of the piperidine ring at N1. The N1-C11 bond length is 1.371 (4) Å. This shows partial double-bond character due to the delocalization of the N-atom lone pair of electrons over the ----N----C==O moiety. The displacement of N1 from the plane defined by C2, C9 and C11 is -0.172(3) Å. As evident from the torsion angles (Table 2), the piperidine ring adopts a flexible twist conformation. The benzoyl group substituted on N1 is in an equatorial orientation; according to Aroney & Le Fèvre (1960), bulky substituents on the N atom of piperidine exist mainly in the equatorial position. The phenyl ring is substituted on C2 in an axial position, the torsion angle C9---N1---C2---C21 being $-86.2(3)^{\circ}$. The torsion angles C12-C11-N1-C9 and O11-C11-N1-C9 are -179.1(3) and $0.5(5)^{\circ}$, respectively, which indicate that the peptide-like group formed by C9, N1, C11, O11 and C12 is planar. The fused cyclohexane ring adopts a chair conformation.



Fig. 1. Molecular structure of (I) showing 50% probability displacement ellipsoids. H atoms are omitted for clarity.

The angle between the least-squares planes through the piperidine and cyclohexane rings is $7.9(1)^{\circ}$. The angle between the least-squares planes of the two phenyl rings is $69.2(1)^{\circ}$.

Experimental

A mixture of 2-phenyldecahydroquinalin-4-one (2.29 g, 0.01 mol) in 20 ml of dry benzene, benzovl chloride (1.4 g, 0.01 mol) and 3 drops of triethylamine was refluxed for 2 h. After the completion of the reaction, the reaction mixture was washed with portions of water $(4 \times 10 \text{ ml})$ and dried over anhydrous sodium sulfate. The excess benzene was distilled off under reduced pressure. Recrystallization from ethanol afforded colourless crystals, m.p. 506 K. The yield was 2.7 g, 80%. The density D_m was measured by flotation in CCl₄ and petroleum ether.

Crystal data

$C_{22}H_{23}NO_2$ $M_r = 333.4$ Orthorhombic $P_{2_12_12_1}$ a = 10.001 (1) Å b = 20.603 (2) Å c = 8.601 (1) Å $V = 1772.2 (3) Å^3$ Z = 4 $D_x = 1.250 \text{ Mg m}^{-3}$ $D_m = 1.260 \text{ Mg m}^{-3}$	Cu $K\alpha$ radiation $\lambda = 1.5418$ Å Cell parameters from 25 reflections $\theta = 14-30^{\circ}$ $\mu = 0.59 \text{ mm}^{-1}$ T = 298 K Plate $0.22 \times 0.18 \times 0.14 \text{ mm}$ Colourless
Data collection	
Enraf-Nonius CAD-4 diffractometer $\omega/2\theta$ scans	$R_{int} = 0.03$ $\theta_{max} = 65^{\circ}$ $h = 0 \rightarrow 11$
Absorption correction: none	$\begin{array}{l} k = 0 \longrightarrow 24 \\ l = 0 \longrightarrow 10 \end{array}$

1746 measured reflections 3 standard reflections 1317 independent reflections frequency: 60 min 1288 observed reflections intensity decay: negligible

Refinement

04 011

NL

C2

 $[I > 3\sigma(I)]$

Refinement on F R = 0.037wR = 0.038S = 1.181288 reflections 226 parameters H-atom parameters not refined

 $w = 1/[\sigma^2(F) + 0.001F^2]$ $(\Delta/\sigma)_{\rm max} = 0.002$ $\Delta \rho_{\rm max} = 0.15 \ {\rm e} \ {\rm \AA}^{-3}$ $\Delta \rho_{\rm min} = -0.19 \ {\rm e} \ {\rm \AA}^{-3}$ Extinction correction: none Atomic scattering factors from SHELX76 (Sheldrick, 1976)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters ($Å^2$)

$$U_{\rm eq} = (1/3) \sum_i \sum_j U_{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j.$$

х	v	Z	U_{eq}
-0.0487 (4)	0.1669 (1)	0.6543 (4)	0.091 (1)
0.3057 (2)	-0.0037(1)	1.0196 (4)	0.065 (1)
0.1024 (2)	0.0435 (1)	0.9901 (3)	0.036 (1)
-0.0356(3)	0.0349 (2)	0.9333 (4)	0.037 (1)

C3	-0.0451 (4)	0.0625 (2)	0.7684 (4)	0.041 (1)
C4	-0.0102 (4)	0.1326 (2)	0.7594 (4)	0.046 (1)
C5	0.1505 (3)	0.2192 (2)	0.8507 (5)	0.049 (1)
C6	0.2263 (4)	0.2457 (2)	0.9897 (6)	0.062 (1)
C7	0.3162 (4)	0.1954 (2)	1.0602 (5)	0.062 (2)
C8	0.2403 (3)	0.1336 (2)	1.0990 (4)	0.048 (1)
C9	0.1703 (3)	0.1059 (2)	0.9554 (4)	0.038 (1)
C10	0.0747 (3)	0.1575 (2)	0.8909 (4)	0.040 (1)
C11	0.1827 (3)	-0.0089(2)	1.0204 (4)	0.045 (1)
C12	0.1204 (3)	-0.0736 (2)	1.0540 (4)	0.043 (1)
C13	0.1805 (4)	-0.1281 (2)	0.9907 (4)	0.054 (1)
C14	0.1260 (5)	-0.1890 (2)	1.0121 (5)	0.068 (2)
C15	0.0131 (5)	-0.1963 (2)	1.1009 (6)	0.070 (2
C16	-0.0436 (5)	-0.1433 (2)	1.1711 (5)	0.068 (2
C17	0.0093 (4)	-0.0822 (2)	1.1475 (5)	0.055 (1
C21	-0.1413 (3)	0.0614 (1)	1.0438 (4)	0.038 (1
C22	-0.1116 (4)	0.0849 (2)	1.1913 (4)	0.047 (1
C23	-0.2121 (5)	0.1085 (2)	1.2875 (5)	0.063 (2
C24	-0.3427 (4)	0.1079 (2)	1.2391 (5)	0.069 (2
C25	-0.3738 (4)	0.0835 (2)	1.0951 (6)	0.066 (2
C26	-0.2747 (3)	0.0603 (2)	0.9979 (5)	0.050 (1

Table 2. Selected geometric parameters (Å, °)

	0	•	
O4—C4	1.210 (5)	N1-C2	1.475 (4)
011—C11	1.235 (4)	N1-C9	1.484 (4)
O4—C4—C3	122.1 (3)	C13—C12—C17	118.2 (4)
C3-C4-C10	115.1 (3)	C2-C21-C26	118.9 (3)
O4-C4-C10	122.8 (4)	C2C21C22	123.1 (3)
C11-C12-C17	124.0 (4)	C22-C21-C26	118.0 (3)
C11C12C13	117.7 (3)		
C2-N1-C11-C12	-22.8 (5)	O4-C4-C10-C9	-148.9 (4)
C2-N1-C9-C10	11.0 (4)	C6—C5—C10—C9	-56.7 (4)
C11-N1-C9-C8	-69.1 (4)	C10C5C6C7	54.0 (5)
C2-N1-C9-C8	133.7 (3)	C5-C6-C7-C8	-53.5 (5)
C9-N1-C2-C3	41.4 (4)	C6—C7—C8—C9	56.6 (4)
N1-C2-C21-C22	-6.7 (5)	C7—C8—C9—C10	-58.2 (4)
N1-C2-C3-C4	-59.2 (4)	N1-C9-C10-C4	-49.4 (4)
C3—C2—C21—C22	-131.7 (4)	C8-C9-C10C5	58.2 (4)
C2-C3-C4-C10	21.9 (5)	N1-C11-C12-C13	139.8 (3)
C3-C4-C10C5	158.3 (3)	011-C11-C12-C13	- 39.8 (5)
C3-C4-C10C9	31.8 (4)		

The structure was solved by direct methods using *SHELXS86* (Sheldrick, 1985). All non-H atoms were refined with anisotropic displacement parameters. All H atoms were obtained from difference Fourier maps and were included in the structure-factor calculations; they were given displacement parameters equal to $1.1U_{eq}$ of their respective carrier atom, but their parameters were not refined (Sheldrick, 1976). The geometrical calculations were performed using *PARST* (Nardelli, 1983).

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates, complete geometry, least-squares-planes data and torsion angles have been deposited with the IUCr (Reference: VJ1020). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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o-Phenylenediammonium Bis(hydrogensulfide)

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Abstract

The title compound, $C_6H_{10}N_2^{2+}.2HS^-$, forms crystals with C2/c symmetry. The anionic HS⁻ groups are located near the NH₃⁺ substituents and are oriented practically perpendicular to the benzene ring.

Comment

The molecule in the unit cell of the title compound, (I), is situated on the twofold axis, which passes through the midpoints of the C(4A)—C(4B) and C(1A)—C(1B) bonds.



The presence of H atoms at every N atom and the C(4A)—N(1A) bond length of 1.457(2) Å indicate protonation of both amino groups. The anionic HS⁻ groups are oriented practically perpendicular to the benzene ring [104 (1)°]. The short interatomic distances N(1A)···S(A) [3.11 (1) Å] and H1(N1A)···S(A) [2.13 (1) Å] confirm the existence of a strong electrostatic interaction between the dication and the anions.